

Structure of Oxolucidine A, a Lycopodium Alkaloid

Motoo Tori,*a Tatsue Shimoji, Shigeru Takaoka, Katsuyuki Nakashima, Masakazu Sono, and William A. Ayer

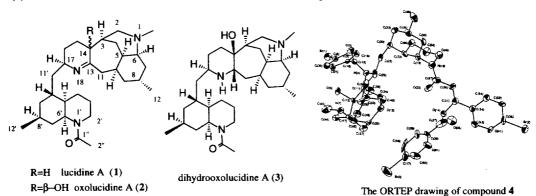
^a Faculty of Pharmaceutical Sciences, Tokushima Bunri University, Yamashiro-cho, Tokushima 770-8514, Japan
^b Department of Chemistry, University of Alberta, Edmonton, Alberta, Canada

Received 7 September 1998; revised 12 October 1998; accepted 30 October 1998

Abstract: A Lycopodium alkaloid, oxolucidine A, was treated with NaBH₄ followed by p-bromobenzoyl chloride to afford a tribenzoate derivative, which was analyzed by X-ray crystallography to establish the stereostructure. The structure of lucidine A was also determined from these results except for the configuration at C-14. © 1998 Elsevier Science Ltd. All rights reserved.

Keywords: Alkaloids; Natural products; NMR; X-Ray crystal structures

Lycopodium plants have long been studied and many alkaloids have been reported thus far [1-7]. Most of the compounds reported have a common formula of $C_{10}N$ or $C_{16}N$ [7], however, among them we have previously reported $C_{30}N_3$ alkaloids [5] named lucidine A (1), lucidine B, oxolucidine A (2), and oxolucidine B. Structures of these alkaloids were highly complicated to solve by simple NMR techniques or simple degradation procedures. Thus, studies in this area are neglected for some time. Recently, huperzine has been isolated from Huperzia serrata (=L. serratum) as a potent inhibitor against the acetylcholine esterase [8]. This prompted us to reinvestigate the chemical constituents of the L. lucidulum extracts. Weak bases were separated using ten-funnel countercurrent distribution as we reported before [6]. Separation of lucidines A (1), B, oxolucidines A (2), and B was carried out by aluminum column chromatography with different activities, followed by reversed-phase HPLC using CH_3CN-H_2O (7:3) as solvent system. Oxolucidines A (2) and B were also obtained from lucidines A (1) and B on exposure to the air [5, 9]. Because there is an



acetamide moiety, its NMR is not so easy to analyze at the room temperature [10, 11]. We now report the full structure of oxolucidine A (2) established by X-ray crystallographic analysis.

Oxolucidine A (2) was converted into dihydrooxolucidine A (3) [12], $C_{30}H_{51}N_{3}O_{2}$, by treatment with NaBH₄ in MeOH. This in turn was treated with *p*-bromobenzoyl chloride and triethylamine in CH₂Cl₂ to afford crystals (m.p. 228-232 °C) [13] from MeOH. The ¹H NMR spectrum of this compound 4 showed the presence of more than two *p*-bromobenzoyl moiety and no acetamide group. The FABMS spectrum showed the multiplet molecular ion peak at m/z 1038, 1036, 1034, and 1032 indicating the presence of the three *p*-bromobenzoyl groups. Because it is not easy to analyze its NMR spectra, X-ray analysis was performed to yield a very unusual structure having two *p*-bromobenzoyl unit at N-acetamide position [14]. This type of structure can be found in the literature [15], although it is unusual to obtain such a compound from simple *p*-bromobenzoylation. Because the η value was +1.081, the absolute configuration of 4 was determined as depicted in the formula. The three chiral centers of 2 at C-17, 5', and 6' positions are the same as those of spirolucidine [6], while different from those of oxolucidine B [5, 6].

The presence of lucidine A and oxolucidine A has been known for a long time, but their structures have been unknown, and this problem has now successfully been solved using X-ray crystallographic analysis. Although the choline esterase inhibitory activity of lucidine A and oxolucidine A has been tested, they showed no activity at all.

Acknowledgements. We thank Dr. M. Tanaka, Miss Y. Kan, and Miss Y. Okamoto, this university, for measurement of the 600 MHz NMR and MS spectra. We express our sincere thanks to Takeda and Co. for biological tests. This work was supported in part by a Grant-in-Aid from the Ministry of Education, Science and Culture, Japan..

References and Notes

- [1] Ayer WA, Berezowsky JA, Law DA. Can. J. Chem. 1963;41:649-657.
- [2] Ayer WA, Altenkirk B, Burnell RH, Moinas M. Can. J. Chem. 1969;47:449-455.
- [3] Ayer WA, Masaki N, Nkunika DS. Can. J. Chem. 1968;46:3631-3642.
- [4] Ayer WA, Altenkirk B. Can. J. Chem. 1969;47:499-502.
- [5] Ayer WA, Browne LM, Nakahara Y, Tori M, Delbaere LTJ. Can. J. Chem. 1979;57:1105-1107.
- [6] Ayer WA, Ball LF, Browne LM, Tori M, Delbaere LTJ, Silverberg A. Can. J. Chem. 1984;62:298-302.
- [7] Ayer WA, Trifonov, LS. Lycopodium Alkaloids. In: The Alkaloids. Academic Press Inc.: 1994;45:233-265.
- [8] Liu JS, Zhu YL, Yu CM, Zhou YZ, Han YY, Wu FW, Qui BF. Can. J. Chem. 1986;64:837-839.
- [9] Cohen LA, Witkop B. J. Am. Chem. Soc. 1955;77:6595-6600.
- [10] Nkunika DS. Ph.D. Thesis, University of Alberta, Edmonton, Alberta, 1967.
- [11] Ball LF. Ph.D. Thesis, University of Alberta, Edmonton, Alberta, 1971.
- [12] 3; $[\alpha]_D^{21}$ +3.9 (c 0.79, CHCl₃); HRMS found m/z 485.3903. Calcd for $C_{30}H_{51}N_3O_2$ 485.3903; MS m/z 485 (M⁺), 467 (base), 452, 424, 410, 291, 273, 259, 248, 235; IR (FTIR) 3400, 2925, 2860, 2775, 1635, 1620, 750 cm⁻¹. ¹³C NMR (150 MHz, CDCl₃) δ 168.9, 168.8, 72.5, 58.6, 58.5, 52.8, 46.9, 43.2, 41.9, 39.9, 38.5, 38.4, 38.3, 36.5, 34.5, 33.8, 33.6, 32.4, 27.1, 26.9, 26.5, 25.8, 25.7, 25.6, 22.9, 22.6, 22.5, 22.3, 21.6, 21.5.
- [13] 4; mp. 228-232°C (from MeOH); $[\alpha]_D^{21}$ +7.3 (c 1.1, CHCl₃); HRMS (FAB) found m/z 1032.2150 [M+H]⁺. Calcd for C₅₁H₆₁O₅N₃Br₃ 1032.2155; MS (FAB) m/z: 1038, 1036, 1034, 1032 [M+H]⁺, 1017, 850, 624, 459, 329, 307, 183 (base), 154, 136; FTIR 3450, 2925, 2860, 2775, 1740, 1625 cm⁻¹; ¹H NMR (600 MHz) (CDCl₃) δ 0.75 (3 H, d, J = 6.32 Hz), 1.24 (3 H, d, J = 7.14 Hz), 1.25 (3 H, s).
- [14] Crystallographic data for 4 have been deposited at the Cambridge Crystallographic Data Centre.
- [15] Kira MA, Zayed AA, Fathy NM. Egypt. J. Chem. 1983;26:253-254.